

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

(1*S*,2*S*,3*S*)-*N*-(2-(4-Chlorophenyl)-1-[*N*-[2-methyl-1-(*N*-methylcarbamoyl)-propyl]carbamoyl]-3-phenylcyclopropyl)benzamideWan-Yun Huang,^a Gui-Fa Su,^{a*} Cheng-Xue Pan,^a Zi-Lu Chen^a and Yong Zhang^b^aCollege of Chemistry and Chemical Engineering, Guangxi Normal University, Guilin 541004, People's Republic of China, and ^bCollege of Chemistry and Chemical Engineering, Suzhou University, Suzhou 215006, People's Republic of China
Correspondence e-mail: edward_su75@163.com

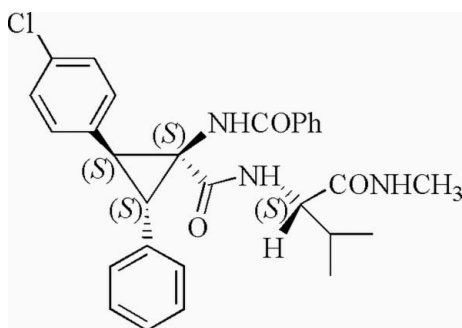
Received 5 June 2007; accepted 4 August 2007

Key indicators: single-crystal X-ray study; $T = 223$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.039; wR factor = 0.087; data-to-parameter ratio = 14.2.

The title compound, $\text{C}_{29}\text{H}_{30}\text{ClN}_3\text{O}_3$, is a conformationally restricted model compound for dipeptides. The diastereomers were separated by column chromatography and the absolute configuration of the phenylalanine cyclopropyl unit of the investigated enantiomer was found to be 2*S*,3*S* by anomalous dispersion. The crystal structure involves intermolecular $\text{N}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For related literature, see: Casanovas *et al.* (2003); Huang *et al.* (2007); Jiménez *et al.* (2005); Royo *et al.* (2005); Su *et al.* (2003).



Experimental

Crystal data

 $\text{C}_{29}\text{H}_{30}\text{ClN}_3\text{O}_3$
 $M_r = 504.01$
Monoclinic, $P2_1$
 $a = 10.8821$ (19) Å
 $b = 9.5688$ (16) Å
 $c = 12.952$ (2) Å
 $\beta = 95.754$ (4)° $V = 1341.9$ (4) Å³
 $Z = 2$
Mo $K\alpha$ radiation
 $\mu = 0.18$ mm⁻¹
 $T = 223$ (2) K
0.50 × 0.36 × 0.30 mm

Data collection

Rigaku Mercury diffractometer
Absorption correction: multi-scan
(Jacobson, 1998)
 $T_{\min} = 0.720$, $T_{\max} = 0.948$ 13162 measured reflections
4678 independent reflections
4334 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.028$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.087$
 $S = 1.09$
4678 reflections
329 parameters
1 restraintH-atom parameters constrained
 $\Delta\rho_{\max} = 0.13$ e Å⁻³
 $\Delta\rho_{\min} = -0.19$ e Å⁻³
Absolute structure: Flack (1983),
with 2067 Friedel pairs
Flack parameter: 0.03 (7)

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|---|--------------|--------------------|-------------|----------------------|
| $\text{N1}-\text{H1}\cdots\text{O3}^i$ | 0.87 | 2.00 | 2.854 (2) | 166 |
| $\text{N2}-\text{H2}\cdots\text{O2}^i$ | 0.87 | 2.17 | 2.912 (2) | 144 |
| $\text{N3}-\text{H3}\cdots\text{O2}^i$ | 0.87 | 2.25 | 3.114 (2) | 173 |
| $\text{C14}-\text{H14B}\cdots\text{O1}^i$ | 0.97 | 2.42 | 3.283 (3) | 148 |

Symmetry code: (i) $-x + 1, y - \frac{1}{2}, -z + 1$.

Data collection: *CrystalClear* (Rigaku/MSK, 2000); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku, 1999); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 2000); software used to prepare material for publication: *SHELXTL*.

The authors thank the Foundation of 100 Young and Middle-Aged Disciplinary Leaders of Guangxi Zhuang Autonomous Region in the 21st Century (grant No. 2004219) and the Foundation of Innovation Projects of Guangxi Graduate Education (grant No. 2006106020703M32).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZL2036).

References

- Casanovas, J., Jiménez, A. I., Cativiela, C., Pérez, J. J. & Alemán, C. (2003). *J. Org. Chem.* **68**, 7088–7091.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Huang, W.-Y., Su, G.-F., Pan, C.-X. & Qin, J.-K. (2007). *Acta Cryst.* **E63**, o2686–o2687.
- Jacobson, R. (1998). Personal communication to Rigaku Corporation, Tokyo, Japan.
- Jiménez, A. I., Ballano, G. & Cativiela, C. (2005). *Angew. Chem. Int. Ed.* **44**, 396–399.
- Rigaku (1999). *CrystalStructure*. Rigaku Corporation, Tokyo, Japan.
- Rigaku/MSK (2000). *CrystalClear*. Rigaku/MSK, The Woodlands, Texas, USA.
- Royo, S., Borggraeve, W. M. D., Peggion, C., Formaggio, F., Crisma, M., Jiménez, A. I., Cativiela, C. & Toniolo, C. (2005). *J. Am. Chem. Soc.* **127**, 2036–2037.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Sheldrick, G. M. (2000). *SHELXTL*. Version 6.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Su, G.-F., Mu, H.-T., Za, D.-M., Zeng, L. M., Cativiela, C., Hammer, R. P. & Yu, K. B. (2003). *Synth. Commun.* **33**, 2873–2884.

supplementary materials

Acta Cryst. (2007). E63, o3743 [doi:10.1107/S1600536807038391]

(1*S*,2*S*,3*S*)-*N*-(2-(4-Chlorophenyl)-1-{*N*-[2-methyl-1-(*N*-methylcarbamoyl)propyl]carbamoyl}-3-phenylcyclopropyl)benzamide

W.-Y. Huang, G.-F. Su, C.-X. Pan, Z.-L. Chen and Y. Zhang

Comment

The synthesis of conformationally restricted model peptides has become one of the research foci in bioorganic and medicinal chemistry in recent years. One common strategy in the design of conformationally restricted model peptides is to incorporate conformationally restricted amino acids into peptide chains to reduce the degrees of freedom of the peptide, which will greatly enhance their selectivity to the receptor, a prerequisite for their successful use as biologically active compounds (Jiménez *et al.*, 2005; Royo *et al.*, 2005; Casanovas *et al.*, 2003).

The title compound (I) (PhCO-*c*₃diAr-Val-NHCH₃) is a conformationally restricted model dipeptide. The absolute configuration of the *c*₃diAr residue was established as (2*S*,3*S*) (Fig. 1). The two substituents of the phenyl and *p*-chloro-phenyl on the three-membered ring are in a *trans* arrangement. The rigidity of the three-membered ring fixes the side chain in a well defined orientation and this disposition is different for each stereoisomer. There is a compromise between the inflexible molecule and its desire to form strong N—H···O hydrogen bonds that leads to a very short H···H contact of 2.134 (1) Å (H29···H12). The crystal packing is stabilized by several N—H···O hydrogen bonds and there are also a several relatively short intermolecular C—H···O hydrogen bonds (Table 1). All these interactions link the molecules into one-dimensional chains and are attributable to stabilize the structure (Fig. 2).

Experimental

0.83 g (2.12 mmol) of 1-benzoylamino-2-(4-chloro-phenyl)-3-phenyl-cyclopropanecarboxylic acid (1), obtained by treatment of 1-benzoylamino-2-(4-chloro-phenyl)-3-phenyl-cyclopropanecarboxylic acid methyl ester with a solution of 2 *M* potassium hydroxide in anhydrous methanol (Su *et al.*, 2003, Huang *et al.*, 2007), was coupled with 2-amino-3-methyl-butyric acid methyl ester hydrochloride (0.36 g, 2.12 mmol) at 254 K by the classic mixed anhydride method using isobutyl chloroformate as the coupling agent. Then the resulting mixture of diastereoisomers (2 and 3 in Figure 3) (0.65 g, white solid) was aminolyzed in a 9.8 *M* methanolic solution of MeNH₂ (15 ml). A careful column chromatography on silica gel (eluent: CHCl₃/AcOEt 5:3, *v/v*) affords the corresponding enantiomerically pure dipeptides PhCO-(2*S*,3*S*)-*c*₃diAr-Val-NHCH₃ (4) and PhCO-(2*R*,3*R*)-*c*₃diAr-Val-NHCH₃ (5) in 36% and 26% yield respectively (Fig. 3). Recrystallization from CH₂Cl₂/CH₃OH (5:1 *v/v*) afforded colourless crystals (m.p. 516–517 K) of PhCO-(2*S*,3*S*)-*c*₃diAr-Val-NHCH₃ (4).

Refinement

The H atoms bound to C atoms were positioned geometrically and included in the refinement in the riding-model approximation, with C—H = 0.94–0.99 Å and with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl and $1.2U_{\text{eq}}(\text{C})$ for all other H atoms.

Figures

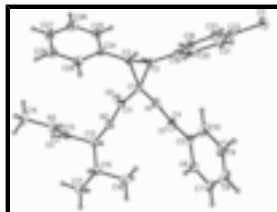


Fig. 1. The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as spheres of arbitrary radii.

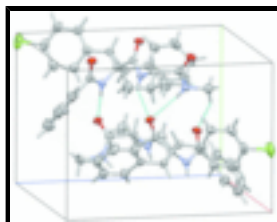


Fig. 2. View of the interactions by H-bonding between adjacent molecules in the unit cell. Displacement ellipsoids drawn at the 30% probability level and H-bonds are indicated as green lines.



Fig. 3. Synthesis of the title compound.

(1*S*,2*S*,3*S*)-*N*-(2-(4-Chlorophenyl)-1-{*N*-[2-methyl-1-(*N*-methylcarbamoyl)propyl]carbamoyl}-3-phenylcyclopropyl)benzamide

Crystal data

$C_{29}H_{30}ClN_3O_3$

$M_r = 504.01$

Monoclinic, $P2_1$

Hall symbol: P 2yb

$a = 10.8821 (19) \text{ \AA}$

$b = 9.5688 (16) \text{ \AA}$

$c = 12.952 (2) \text{ \AA}$

$\beta = 95.754 (4)^\circ$

$V = 1341.9 (4) \text{ \AA}^3$

$Z = 2$

$F_{000} = 532$

$D_x = 1.247 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

$\lambda = 0.71070 \text{ \AA}$

Cell parameters from 5213 reflections

$\theta = 3.2\text{--}25.3^\circ$

$\mu = 0.18 \text{ mm}^{-1}$

$T = 223 (2) \text{ K}$

Block, colourless

$0.50 \times 0.36 \times 0.30 \text{ mm}$

Data collection

Rigaku Mercury
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 223(2) \text{ K}$

ω scans

Absorption correction: multi-scan
(Jacobson, 1998)

4678 independent reflections

4334 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.028$

$\theta_{\text{max}} = 25.3^\circ$

$\theta_{\text{min}} = 3.2^\circ$

$h = -12 \rightarrow 13$

$T_{\min} = 0.720$, $T_{\max} = 0.948$
13162 measured reflections

$k = -11 \rightarrow 11$
 $l = -15 \rightarrow 15$

Refinement

Refinement on F^2

Hydrogen site location: inferred from neighbouring sites

Least-squares matrix: full

H-atom parameters constrained

$R[F^2 > 2\sigma(F^2)] = 0.039$

$$w = 1/[\sigma^2(F_o^2) + (0.0394P)^2 + 0.1618P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$wR(F^2) = 0.087$

$(\Delta/\sigma)_{\max} < 0.001$

$S = 1.09$

$\Delta\rho_{\max} = 0.13 \text{ e } \text{\AA}^{-3}$

4678 reflections

$\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$

329 parameters

Extinction correction: none

1 restraint

Absolute structure: Flack (1983), with 2067 Friedel pairs

Primary atom site location: structure-invariant direct methods

Flack parameter: 0.03 (7)

Secondary atom site location: difference Fourier map

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|--------------|--------------|----------------------------------|
| Cl1 | -0.04548 (6) | 0.93271 (10) | 0.97899 (5) | 0.0693 (2) |
| O1 | 0.51546 (14) | 0.93017 (16) | 0.76039 (11) | 0.0401 (4) |
| O2 | 0.44949 (13) | 1.03295 (14) | 0.51521 (12) | 0.0343 (3) |
| O3 | 0.69537 (14) | 0.97648 (15) | 0.32163 (12) | 0.0387 (4) |
| N1 | 0.40924 (14) | 0.75020 (17) | 0.68410 (12) | 0.0281 (4) |
| H1 | 0.3900 | 0.6620 | 0.6849 | 0.034* |
| N2 | 0.55739 (14) | 0.83074 (16) | 0.53124 (13) | 0.0276 (4) |
| H2 | 0.5547 | 0.7427 | 0.5478 | 0.033* |
| N3 | 0.61830 (17) | 0.75935 (18) | 0.32730 (13) | 0.0367 (4) |
| H3 | 0.5969 | 0.6912 | 0.3662 | 0.044* |
| C1 | 0.36200 (17) | 0.83804 (19) | 0.59967 (15) | 0.0271 (4) |
| C2 | 0.23957 (17) | 0.9144 (2) | 0.60451 (15) | 0.0308 (4) |
| H2B | 0.2357 | 1.0055 | 0.5681 | 0.037* |

supplementary materials

| | | | | |
|------|--------------|------------|--------------|------------|
| C3 | 0.24678 (18) | 0.7912 (2) | 0.53285 (15) | 0.0297 (4) |
| H3B | 0.2090 | 0.7044 | 0.5568 | 0.036* |
| C4 | 0.48463 (19) | 0.8075 (2) | 0.76281 (16) | 0.0319 (5) |
| C5 | 0.5331 (2) | 0.7109 (2) | 0.84811 (15) | 0.0344 (5) |
| C6 | 0.6533 (2) | 0.7319 (3) | 0.89339 (17) | 0.0434 (6) |
| H6 | 0.7002 | 0.8065 | 0.8710 | 0.052* |
| C7 | 0.7041 (3) | 0.6441 (3) | 0.97087 (19) | 0.0553 (7) |
| H7 | 0.7860 | 0.6574 | 0.9996 | 0.066* |
| C8 | 0.6352 (3) | 0.5375 (3) | 1.00590 (19) | 0.0567 (7) |
| H8 | 0.6701 | 0.4773 | 1.0582 | 0.068* |
| C9 | 0.5153 (3) | 0.5188 (3) | 0.9647 (2) | 0.0634 (8) |
| H9 | 0.4671 | 0.4477 | 0.9905 | 0.076* |
| C10 | 0.4646 (3) | 0.6050 (3) | 0.88447 (18) | 0.0510 (7) |
| H10 | 0.3831 | 0.5903 | 0.8553 | 0.061* |
| C11 | 0.46021 (17) | 0.9113 (2) | 0.54537 (14) | 0.0265 (4) |
| C12 | 0.66756 (17) | 0.8836 (2) | 0.48946 (15) | 0.0290 (4) |
| H12 | 0.6767 | 0.9831 | 0.5097 | 0.035* |
| C13 | 0.65934 (18) | 0.8770 (2) | 0.37172 (16) | 0.0308 (5) |
| C14 | 0.6078 (3) | 0.7401 (3) | 0.21541 (17) | 0.0482 (6) |
| H14A | 0.5414 | 0.7981 | 0.1835 | 0.072* |
| H14B | 0.5901 | 0.6428 | 0.1990 | 0.072* |
| H14C | 0.6848 | 0.7667 | 0.1890 | 0.072* |
| C15 | 0.78250 (18) | 0.8054 (2) | 0.53901 (17) | 0.0370 (5) |
| H15 | 0.7725 | 0.7052 | 0.5213 | 0.044* |
| C16 | 0.7953 (2) | 0.8184 (3) | 0.65662 (19) | 0.0559 (7) |
| H16A | 0.8710 | 0.7732 | 0.6851 | 0.084* |
| H16B | 0.7253 | 0.7740 | 0.6840 | 0.084* |
| H16C | 0.7978 | 0.9165 | 0.6758 | 0.084* |
| C17 | 0.8986 (2) | 0.8572 (4) | 0.4957 (2) | 0.0622 (8) |
| H17A | 0.8876 | 0.8524 | 0.4205 | 0.093* |
| H17B | 0.9680 | 0.7992 | 0.5216 | 0.093* |
| H17C | 0.9144 | 0.9532 | 0.5171 | 0.093* |
| C18 | 0.16989 (17) | 0.9123 (2) | 0.69795 (15) | 0.0325 (5) |
| C19 | 0.1195 (2) | 1.0362 (2) | 0.72906 (18) | 0.0394 (5) |
| H19 | 0.1309 | 1.1181 | 0.6910 | 0.047* |
| C20 | 0.0527 (2) | 1.0435 (3) | 0.81444 (19) | 0.0465 (6) |
| H20 | 0.0181 | 1.1286 | 0.8335 | 0.056* |
| C21 | 0.03798 (19) | 0.9244 (3) | 0.87064 (16) | 0.0435 (5) |
| C22 | 0.0854 (2) | 0.7992 (3) | 0.84255 (18) | 0.0434 (6) |
| H22 | 0.0742 | 0.7183 | 0.8817 | 0.052* |
| C23 | 0.1505 (2) | 0.7927 (2) | 0.75545 (17) | 0.0381 (5) |
| H23 | 0.1817 | 0.7064 | 0.7352 | 0.046* |
| C24 | 0.23451 (18) | 0.8098 (2) | 0.41708 (16) | 0.0308 (4) |
| C25 | 0.1599 (2) | 0.9128 (3) | 0.36832 (17) | 0.0420 (5) |
| H25 | 0.1180 | 0.9754 | 0.4084 | 0.050* |
| C26 | 0.1466 (2) | 0.9245 (3) | 0.26140 (19) | 0.0556 (7) |
| H26 | 0.0961 | 0.9951 | 0.2295 | 0.067* |
| C27 | 0.2066 (2) | 0.8335 (3) | 0.20145 (19) | 0.0543 (7) |
| H27 | 0.1963 | 0.8411 | 0.1287 | 0.065* |

| | | | | |
|-----|------------|------------|--------------|------------|
| C28 | 0.2814 (2) | 0.7317 (3) | 0.24767 (17) | 0.0473 (6) |
| H28 | 0.3229 | 0.6696 | 0.2068 | 0.057* |
| C29 | 0.2959 (2) | 0.7204 (2) | 0.35476 (16) | 0.0378 (5) |
| H29 | 0.3483 | 0.6510 | 0.3860 | 0.045* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C11 | 0.0616 (4) | 0.1030 (6) | 0.0466 (4) | -0.0071 (4) | 0.0225 (3) | -0.0152 (4) |
| O1 | 0.0530 (9) | 0.0268 (8) | 0.0392 (8) | -0.0032 (7) | -0.0017 (7) | -0.0039 (7) |
| O2 | 0.0374 (8) | 0.0221 (8) | 0.0439 (9) | 0.0009 (6) | 0.0059 (7) | 0.0054 (6) |
| O3 | 0.0479 (9) | 0.0266 (8) | 0.0430 (9) | -0.0031 (7) | 0.0117 (7) | 0.0034 (6) |
| N1 | 0.0325 (9) | 0.0230 (9) | 0.0283 (9) | -0.0006 (7) | 0.0012 (7) | 0.0025 (7) |
| N2 | 0.0297 (8) | 0.0196 (8) | 0.0342 (9) | 0.0003 (7) | 0.0072 (7) | 0.0005 (7) |
| N3 | 0.0503 (11) | 0.0292 (10) | 0.0313 (9) | -0.0060 (8) | 0.0071 (8) | -0.0007 (8) |
| C1 | 0.0307 (10) | 0.0229 (10) | 0.0277 (10) | 0.0016 (8) | 0.0024 (8) | 0.0013 (8) |
| C2 | 0.0285 (10) | 0.0289 (11) | 0.0349 (11) | 0.0023 (9) | 0.0031 (8) | 0.0039 (9) |
| C3 | 0.0278 (10) | 0.0295 (11) | 0.0314 (11) | -0.0040 (8) | 0.0020 (8) | 0.0028 (9) |
| C4 | 0.0355 (11) | 0.0307 (12) | 0.0298 (11) | 0.0021 (9) | 0.0052 (9) | -0.0040 (9) |
| C5 | 0.0473 (13) | 0.0316 (11) | 0.0241 (10) | 0.0022 (10) | 0.0027 (9) | -0.0018 (9) |
| C6 | 0.0494 (14) | 0.0499 (14) | 0.0302 (12) | 0.0028 (11) | 0.0000 (10) | -0.0003 (11) |
| C7 | 0.0597 (16) | 0.073 (2) | 0.0304 (13) | 0.0187 (15) | -0.0062 (12) | 0.0018 (12) |
| C8 | 0.085 (2) | 0.0533 (16) | 0.0292 (12) | 0.0142 (15) | -0.0068 (13) | 0.0020 (12) |
| C9 | 0.099 (2) | 0.0501 (16) | 0.0384 (14) | -0.0173 (15) | -0.0051 (15) | 0.0113 (12) |
| C10 | 0.0643 (17) | 0.0509 (15) | 0.0352 (13) | -0.0126 (13) | -0.0084 (12) | 0.0073 (12) |
| C11 | 0.0285 (10) | 0.0242 (11) | 0.0260 (10) | -0.0022 (8) | -0.0010 (8) | -0.0018 (8) |
| C12 | 0.0283 (10) | 0.0241 (10) | 0.0352 (11) | -0.0021 (8) | 0.0059 (9) | -0.0019 (8) |
| C13 | 0.0280 (10) | 0.0271 (11) | 0.0377 (11) | 0.0024 (8) | 0.0061 (9) | -0.0017 (9) |
| C14 | 0.0708 (16) | 0.0410 (14) | 0.0331 (12) | -0.0061 (13) | 0.0069 (11) | -0.0040 (10) |
| C15 | 0.0301 (10) | 0.0408 (13) | 0.0397 (12) | 0.0033 (9) | 0.0007 (9) | -0.0066 (10) |
| C16 | 0.0470 (14) | 0.076 (2) | 0.0427 (14) | 0.0094 (14) | -0.0037 (11) | 0.0017 (13) |
| C17 | 0.0314 (12) | 0.093 (2) | 0.0618 (17) | -0.0015 (13) | 0.0043 (12) | 0.0001 (16) |
| C18 | 0.0247 (9) | 0.0368 (12) | 0.0353 (11) | -0.0014 (9) | 0.0000 (8) | 0.0008 (10) |
| C19 | 0.0376 (12) | 0.0381 (12) | 0.0424 (13) | 0.0015 (10) | 0.0041 (10) | -0.0028 (10) |
| C20 | 0.0414 (13) | 0.0520 (15) | 0.0464 (14) | 0.0052 (12) | 0.0053 (11) | -0.0135 (12) |
| C21 | 0.0331 (11) | 0.0657 (16) | 0.0318 (11) | -0.0050 (12) | 0.0044 (9) | -0.0103 (12) |
| C22 | 0.0380 (12) | 0.0546 (15) | 0.0380 (13) | -0.0050 (11) | 0.0064 (10) | 0.0074 (11) |
| C23 | 0.0353 (11) | 0.0380 (13) | 0.0413 (12) | 0.0009 (10) | 0.0057 (10) | 0.0020 (10) |
| C24 | 0.0275 (10) | 0.0315 (11) | 0.0329 (11) | -0.0064 (9) | 0.0013 (8) | 0.0056 (9) |
| C25 | 0.0404 (12) | 0.0446 (14) | 0.0397 (12) | 0.0048 (11) | -0.0023 (10) | 0.0077 (11) |
| C26 | 0.0577 (15) | 0.0586 (16) | 0.0473 (14) | 0.0079 (14) | -0.0106 (12) | 0.0181 (14) |
| C27 | 0.0585 (15) | 0.0732 (19) | 0.0300 (12) | -0.0068 (15) | -0.0016 (11) | 0.0097 (12) |
| C28 | 0.0528 (14) | 0.0550 (15) | 0.0344 (13) | -0.0027 (13) | 0.0059 (11) | 0.0005 (11) |
| C29 | 0.0398 (12) | 0.0373 (12) | 0.0357 (12) | -0.0004 (10) | 0.0008 (10) | 0.0030 (10) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|---------|-----------|----------|--------|
| C11—C21 | 1.748 (2) | C12—H12 | 0.9900 |
| O1—C4 | 1.222 (3) | C14—H14A | 0.9700 |

supplementary materials

| | | | |
|------------|-------------|---------------|-------------|
| O2—C11 | 1.230 (2) | C14—H14B | 0.9700 |
| O3—C13 | 1.238 (2) | C14—H14C | 0.9700 |
| N1—C4 | 1.358 (3) | C15—C17 | 1.516 (3) |
| N1—C1 | 1.433 (2) | C15—C16 | 1.521 (3) |
| N1—H1 | 0.8700 | C15—H15 | 0.9900 |
| N2—C11 | 1.336 (2) | C16—H16A | 0.9700 |
| N2—C12 | 1.455 (2) | C16—H16B | 0.9700 |
| N2—H2 | 0.8700 | C16—H16C | 0.9700 |
| N3—C13 | 1.321 (3) | C17—H17A | 0.9700 |
| N3—C14 | 1.454 (3) | C17—H17B | 0.9700 |
| N3—H3 | 0.8700 | C17—H17C | 0.9700 |
| C1—C11 | 1.510 (3) | C18—C19 | 1.384 (3) |
| C1—C3 | 1.518 (3) | C18—C23 | 1.393 (3) |
| C1—C2 | 1.526 (3) | C19—C20 | 1.385 (3) |
| C2—C18 | 1.491 (3) | C19—H19 | 0.9400 |
| C2—C3 | 1.507 (3) | C20—C21 | 1.370 (4) |
| C2—H2B | 0.9900 | C20—H20 | 0.9400 |
| C3—C24 | 1.503 (3) | C21—C22 | 1.369 (4) |
| C3—H3B | 0.9900 | C22—C23 | 1.392 (3) |
| C4—C5 | 1.496 (3) | C22—H22 | 0.9400 |
| C5—C10 | 1.369 (3) | C23—H23 | 0.9400 |
| C5—C6 | 1.393 (3) | C24—C25 | 1.389 (3) |
| C6—C7 | 1.381 (3) | C24—C29 | 1.392 (3) |
| C6—H6 | 0.9400 | C25—C26 | 1.382 (3) |
| C7—C8 | 1.369 (4) | C25—H25 | 0.9400 |
| C7—H7 | 0.9400 | C26—C27 | 1.375 (4) |
| C8—C9 | 1.371 (4) | C26—H26 | 0.9400 |
| C8—H8 | 0.9400 | C27—C28 | 1.368 (4) |
| C9—C10 | 1.396 (3) | C27—H27 | 0.9400 |
| C9—H9 | 0.9400 | C28—C29 | 1.384 (3) |
| C10—H10 | 0.9400 | C28—H28 | 0.9400 |
| C12—C13 | 1.520 (3) | C29—H29 | 0.9400 |
| C12—C15 | 1.541 (3) | | |
| C4—N1—C1 | 118.66 (16) | N3—C14—H14A | 109.5 |
| C4—N1—H1 | 120.7 | N3—C14—H14B | 109.5 |
| C1—N1—H1 | 120.7 | H14A—C14—H14B | 109.5 |
| C11—N2—C12 | 122.92 (16) | N3—C14—H14C | 109.5 |
| C11—N2—H2 | 118.5 | H14A—C14—H14C | 109.5 |
| C12—N2—H2 | 118.5 | H14B—C14—H14C | 109.5 |
| C13—N3—C14 | 122.09 (19) | C17—C15—C16 | 110.4 (2) |
| C13—N3—H3 | 119.0 | C17—C15—C12 | 111.0 (2) |
| C14—N3—H3 | 119.0 | C16—C15—C12 | 111.50 (18) |
| N1—C1—C11 | 114.29 (16) | C17—C15—H15 | 107.9 |
| N1—C1—C3 | 118.32 (16) | C16—C15—H15 | 107.9 |
| C11—C1—C3 | 117.03 (16) | C12—C15—H15 | 107.9 |
| N1—C1—C2 | 119.81 (16) | C15—C16—H16A | 109.5 |
| C11—C1—C2 | 117.30 (16) | C15—C16—H16B | 109.5 |
| C3—C1—C2 | 59.35 (13) | H16A—C16—H16B | 109.5 |
| C18—C2—C3 | 123.47 (18) | C15—C16—H16C | 109.5 |

| | | | |
|---------------|--------------|----------------|--------------|
| C18—C2—C1 | 122.95 (17) | H16A—C16—H16C | 109.5 |
| C3—C2—C1 | 60.06 (13) | H16B—C16—H16C | 109.5 |
| C18—C2—H2B | 113.4 | C15—C17—H17A | 109.5 |
| C3—C2—H2B | 113.4 | C15—C17—H17B | 109.5 |
| C1—C2—H2B | 113.4 | H17A—C17—H17B | 109.5 |
| C24—C3—C2 | 121.14 (18) | C15—C17—H17C | 109.5 |
| C24—C3—C1 | 121.31 (17) | H17A—C17—H17C | 109.5 |
| C2—C3—C1 | 60.59 (13) | H17B—C17—H17C | 109.5 |
| C24—C3—H3B | 114.5 | C19—C18—C23 | 117.56 (19) |
| C2—C3—H3B | 114.5 | C19—C18—C2 | 118.3 (2) |
| C1—C3—H3B | 114.5 | C23—C18—C2 | 124.2 (2) |
| O1—C4—N1 | 120.99 (19) | C18—C19—C20 | 122.1 (2) |
| O1—C4—C5 | 122.35 (19) | C18—C19—H19 | 119.0 |
| N1—C4—C5 | 116.56 (18) | C20—C19—H19 | 119.0 |
| C10—C5—C6 | 118.9 (2) | C21—C20—C19 | 118.7 (2) |
| C10—C5—C4 | 123.2 (2) | C21—C20—H20 | 120.6 |
| C6—C5—C4 | 117.9 (2) | C19—C20—H20 | 120.6 |
| C7—C6—C5 | 120.5 (2) | C22—C21—C20 | 121.3 (2) |
| C7—C6—H6 | 119.7 | C22—C21—C11 | 119.6 (2) |
| C5—C6—H6 | 119.7 | C20—C21—C11 | 119.1 (2) |
| C8—C7—C6 | 120.1 (3) | C21—C22—C23 | 119.4 (2) |
| C8—C7—H7 | 119.9 | C21—C22—H22 | 120.3 |
| C6—C7—H7 | 119.9 | C23—C22—H22 | 120.3 |
| C7—C8—C9 | 120.0 (2) | C22—C23—C18 | 120.9 (2) |
| C7—C8—H8 | 120.0 | C22—C23—H23 | 119.6 |
| C9—C8—H8 | 120.0 | C18—C23—H23 | 119.6 |
| C8—C9—C10 | 120.1 (3) | C25—C24—C29 | 117.74 (19) |
| C8—C9—H9 | 119.9 | C25—C24—C3 | 121.83 (19) |
| C10—C9—H9 | 119.9 | C29—C24—C3 | 120.41 (18) |
| C5—C10—C9 | 120.4 (2) | C26—C25—C24 | 120.7 (2) |
| C5—C10—H10 | 119.8 | C26—C25—H25 | 119.6 |
| C9—C10—H10 | 119.8 | C24—C25—H25 | 119.6 |
| O2—C11—N2 | 123.64 (18) | C27—C26—C25 | 120.4 (2) |
| O2—C11—C1 | 122.59 (18) | C27—C26—H26 | 119.8 |
| N2—C11—C1 | 113.75 (16) | C25—C26—H26 | 119.8 |
| N2—C12—C13 | 113.02 (16) | C28—C27—C26 | 120.0 (2) |
| N2—C12—C15 | 109.95 (16) | C28—C27—H27 | 120.0 |
| C13—C12—C15 | 111.18 (16) | C26—C27—H27 | 120.0 |
| N2—C12—H12 | 107.5 | C27—C28—C29 | 119.8 (2) |
| C13—C12—H12 | 107.5 | C27—C28—H28 | 120.1 |
| C15—C12—H12 | 107.5 | C29—C28—H28 | 120.1 |
| O3—C13—N3 | 122.54 (19) | C28—C29—C24 | 121.3 (2) |
| O3—C13—C12 | 120.27 (18) | C28—C29—H29 | 119.3 |
| N3—C13—C12 | 117.11 (18) | C24—C29—H29 | 119.3 |
| C4—N1—C1—C11 | 56.8 (2) | C11—N2—C12—C13 | -88.6 (2) |
| C4—N1—C1—C3 | -159.11 (18) | C11—N2—C12—C15 | 146.52 (18) |
| C4—N1—C1—C2 | -90.1 (2) | C14—N3—C13—O3 | -2.5 (3) |
| N1—C1—C2—C18 | 5.4 (3) | C14—N3—C13—C12 | -179.27 (19) |
| C11—C1—C2—C18 | -140.58 (19) | N2—C12—C13—O3 | 137.01 (18) |

supplementary materials

| | | | |
|---------------|--------------|-----------------|--------------|
| C3—C1—C2—C18 | 112.6 (2) | C15—C12—C13—O3 | -98.8 (2) |
| N1—C1—C2—C3 | -107.22 (19) | N2—C12—C13—N3 | -46.2 (2) |
| C11—C1—C2—C3 | 106.77 (19) | C15—C12—C13—N3 | 78.0 (2) |
| C18—C2—C3—C24 | 137.30 (19) | N2—C12—C15—C17 | 179.2 (2) |
| C1—C2—C3—C24 | -110.9 (2) | C13—C12—C15—C17 | 53.3 (2) |
| C18—C2—C3—C1 | -111.8 (2) | N2—C12—C15—C16 | -57.2 (2) |
| N1—C1—C3—C24 | -139.70 (19) | C13—C12—C15—C16 | 176.83 (19) |
| C11—C1—C3—C24 | 3.4 (3) | C3—C2—C18—C19 | -149.82 (19) |
| C2—C1—C3—C24 | 110.6 (2) | C1—C2—C18—C19 | 136.7 (2) |
| N1—C1—C3—C2 | 109.70 (19) | C3—C2—C18—C23 | 29.1 (3) |
| C11—C1—C3—C2 | -107.23 (19) | C1—C2—C18—C23 | -44.3 (3) |
| C1—N1—C4—O1 | -2.6 (3) | C23—C18—C19—C20 | 0.5 (3) |
| C1—N1—C4—C5 | -178.99 (17) | C2—C18—C19—C20 | 179.5 (2) |
| O1—C4—C5—C10 | 146.1 (2) | C18—C19—C20—C21 | 1.1 (3) |
| N1—C4—C5—C10 | -37.5 (3) | C19—C20—C21—C22 | -1.5 (3) |
| O1—C4—C5—C6 | -32.8 (3) | C19—C20—C21—C11 | 179.66 (17) |
| N1—C4—C5—C6 | 143.5 (2) | C20—C21—C22—C23 | 0.3 (3) |
| C10—C5—C6—C7 | 2.5 (3) | C11—C21—C22—C23 | 179.18 (17) |
| C4—C5—C6—C7 | -178.5 (2) | C21—C22—C23—C18 | 1.3 (3) |
| C5—C6—C7—C8 | -1.8 (4) | C19—C18—C23—C22 | -1.7 (3) |
| C6—C7—C8—C9 | -0.6 (4) | C2—C18—C23—C22 | 179.35 (19) |
| C7—C8—C9—C10 | 2.3 (4) | C2—C3—C24—C25 | -32.7 (3) |
| C6—C5—C10—C9 | -0.8 (4) | C1—C3—C24—C25 | -105.0 (2) |
| C4—C5—C10—C9 | -179.7 (2) | C2—C3—C24—C29 | 149.11 (19) |
| C8—C9—C10—C5 | -1.6 (4) | C1—C3—C24—C29 | 76.8 (3) |
| C12—N2—C11—O2 | 7.5 (3) | C29—C24—C25—C26 | 0.7 (3) |
| C12—N2—C11—C1 | -174.07 (16) | C3—C24—C25—C26 | -177.6 (2) |
| N1—C1—C11—O2 | -140.36 (18) | C24—C25—C26—C27 | 0.4 (4) |
| C3—C1—C11—O2 | 75.1 (2) | C25—C26—C27—C28 | -0.9 (4) |
| C2—C1—C11—O2 | 7.5 (3) | C26—C27—C28—C29 | 0.3 (4) |
| N1—C1—C11—N2 | 41.2 (2) | C27—C28—C29—C24 | 0.8 (3) |
| C3—C1—C11—N2 | -103.4 (2) | C25—C24—C29—C28 | -1.3 (3) |
| C2—C1—C11—N2 | -170.97 (16) | C3—C24—C29—C28 | 177.0 (2) |

Hydrogen-bond geometry (\AA , $^\circ$)

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-----------------------------------|-------|-------------|-------------|---------------|
| N1—H1 \cdots O3 ⁱ | 0.87 | 2.00 | 2.854 (2) | 166 |
| N2—H2 \cdots O2 ⁱ | 0.87 | 2.17 | 2.912 (2) | 144 |
| N3—H3 \cdots O2 ⁱ | 0.87 | 2.25 | 3.114 (2) | 173 |
| C14—H14B \cdots O1 ⁱ | 0.97 | 2.42 | 3.283 (3) | 148 |

Symmetry codes: (i) $-x+1, y-1/2, -z+1$.

Fig. 1

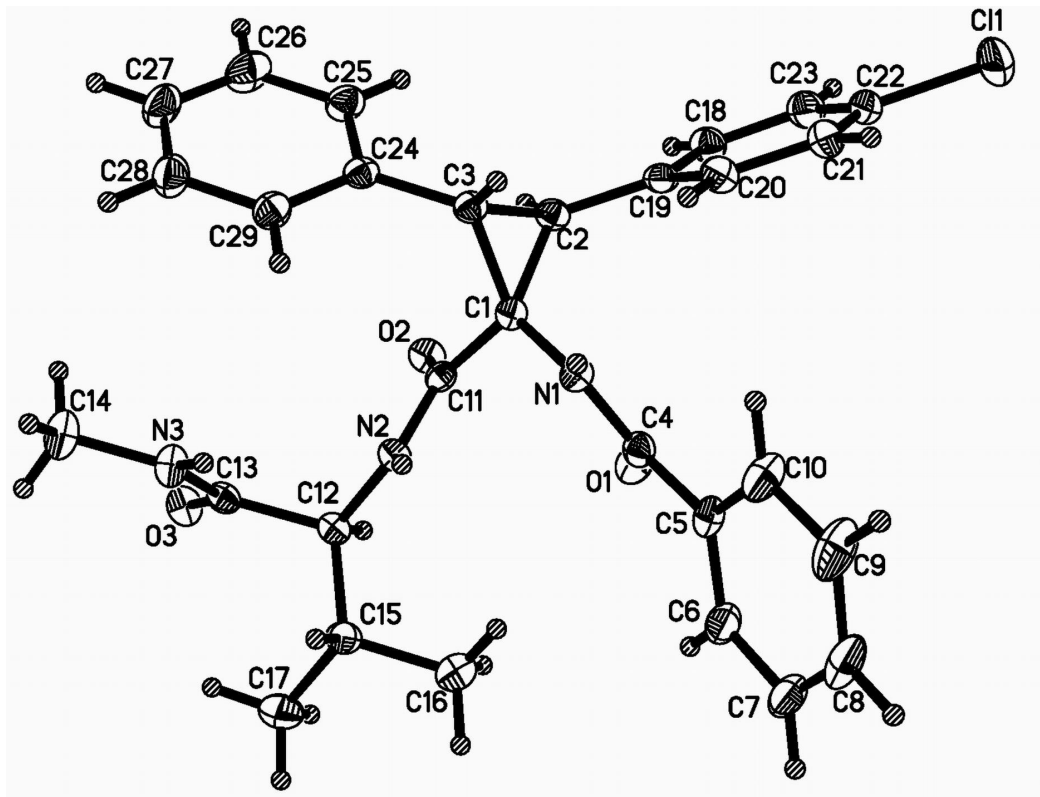


Fig. 2

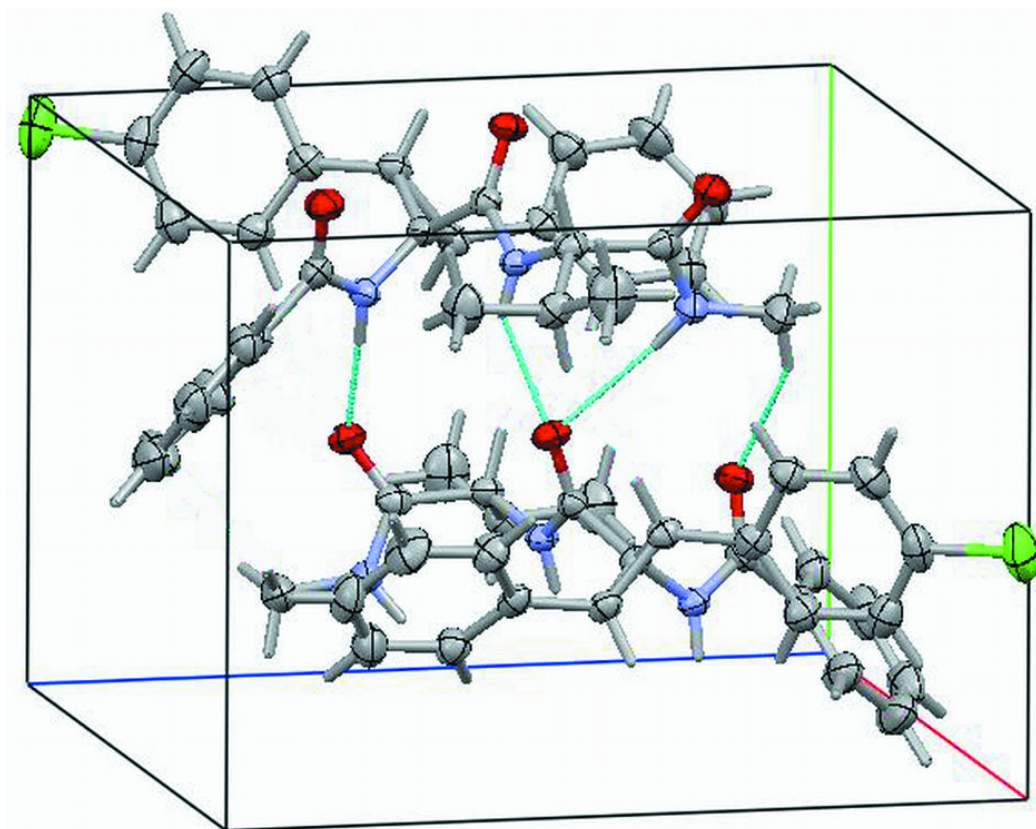


Fig. 3

